

Pathways to the Oxidation of Copper

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Motivation and Accomplishments

Understanding oxidation leads to a better control of:

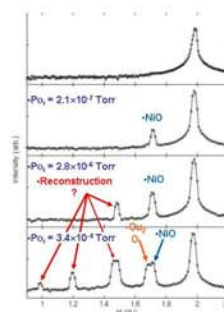
- Corrosion
- Heterogeneous catalysis
- Growth of oxide layers for devices
- Performance of nanoscale materials

Our studies of Cu(001) oxidation:

- Find c(2x2) structure over a wide of (pO_2 , T)
- Explain conflicting results in the literature
- Demonstrate that latency is connected with saturation of this reconstruction

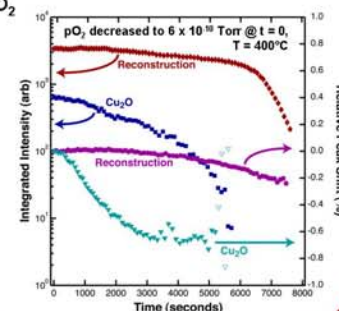
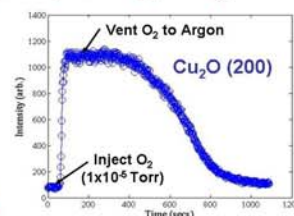
Future Directions

- Alloying causes more complex oxidation behavior (e.g. oxidation of Cu-Ni alloys is very different than that of pure Cu)
- Studies are also being extended to oxide-supported metal and alloy islands (relevant to redox catalysis)
- X-ray scattering studies will be complemented by theory and *in situ* TEM, ESEM, and spectroscopy



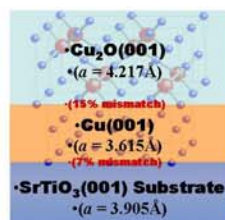
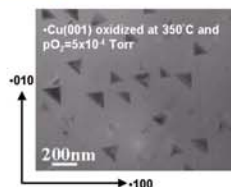
Oxidation and Reduction are Reversible

Cu₂O can be fully reduced by lower pO_2 (below). The c(2x2) reconstruction (right) persists until Cu₂O is gone (oxide phase supplies O₂)



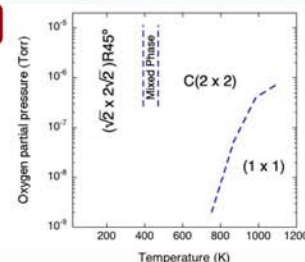
Background Information

- Initial formation of oxide is through islands
- Oxygen induces a reconstruction of the Cu (001) surface, usually $2\sqrt{2} \times \sqrt{2} R45^\circ$
- Theory predicts sub-surface oxygen plays an important role
- Impurities create problems but can be minimized by thin film growth
- We produce epitaxial films on SrTiO₃ substrates by ex-situ evaporation under UHV conditions
- Oxide formation is studied using *in situ* x-ray scattering under quasi-equilibrium conditions



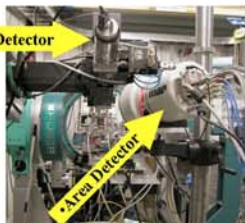
Surface Phase Diagram

- Found an ordered surface structure with c(2x2) symmetry
- c(2x2) is stable over a wide range of T and pO_2 (resolved literature controversy).
- Does surface structure modify oxidation behavior?

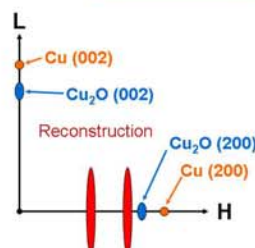


Materials Processing System

- Precisely controlled atmosphere (Ar, O₂, CO, CO₂, H₂) with base pressure <10⁻⁷ Torr
- Temperatures between RT and 1000°C
- Two stage residual gas analysis
- Mounts on standard diffractometer



In Situ X-Ray Measurements



While varying pO_2 , monitor:

- Adsorption (CTRs)
- Cu strain state (●)
- Surface reconstructions (●)
- Cu₂O nucleation & growth (●)
- Cu-Cu₂O phase equilibria (intensity of ● increase or decrease?)

